Mass Spectrometry and its Applications to Organic Chemistry. By J. H. Beynon, Head of the Physics Section, I.C.I. Dyestuffs Division, Manchester (England). D. Van Nostrand Company, Inc., 120 Alexander Street, Princeton, New Jersey. 1960. 'xii + 640 pp. 17.5 × 25 cm. Price, \$24.00.

The bases of the analytical chemistry of complex mixtures have been and continue to be separation and characteriza-For obvious reasons the most used analytical methtion. ods have always been those in which the necessary characterization is more or less automatically coupled to the separation method. The ideal analytical method is one that completely separates the components of a mixture, and simultaneously provides a unique characterization of the sepa-rated components. The reader of the volume under review cannot help but finish his reading with the impression that mass spectrometry as practiced in the laboratory of the author, J. H. Beynon, is a very close approximation to the ideal analytical method. The appearance at this time of a volume that can lead its readers to such a conclusion is most opportune, since it has recently become possible for any laboratory with a sufficiently large annual capital budget to acquire equipment that is equal in potential performance to that employed by Beynon.

Two classes of reader will find this book of particular value. These are, (1) organic chemists who would like to learn whether or not high resolution mass spectrometry will help them in the solution of their analytical and/or structure problems, and (2) analytical spectroscopists who have or are about to have the responsibility for interpreting high resolution mass spectra of complex organic compounds and their mixtures. Beynon's forté, as indicated in the preface to the volume and by the important papers he has contributed to the analytical chemical literature, is qualitative organic analysis by means of high resolution mass spectrometry, *i.e.*, the determination of the exact empirical formulas of substances through precise determination of nuolecular weights. At least 90% of this volume may be viewed as an introductory manual to the art which Beynon has fathered, and/or material that Beynon believes to be pertinent to the practice of this art. The other 10% of the book is a deliberately brief review of some of the other aspects, largely of an analytical chemical nature, of mass spectrometry.

The reviewer intends the implication of the preceding paragraph, to wit, that Beynon's book will be of little interest or use to the chemist (physicist or biologist) who is interested in the state of development of mass spectrometry in the large. Beynon and the publishers are to be congratulated for choosing a relatively unambiguous title for their volume, although greater precision in naming the book would have been achieved by preceding "chemistry" with "analytical" in the title.

With respect to the actual structure and content of the book, the first six chapters (240 pages) describe mass spectrometers and their auxiliary laboratory appurtenances and the measurement of the intensive and extensive aspects of mass spectra. The seventh chapter (48 pages) is a general description of the nature of the mass spectra of molecules that are generated by single electron impact. The eighth and ninth chapters (130 pages), that constitute the heart of the book, are devoted to qualitative analysis by mass spectrometry and the nature of the mass spectra of various classes of substances. As indicated above, there is a tenth chapter (60 pages) that summarizes other applications of mass spectroscopist engaged in interpreting high resolution mass spectra of organic compounds. The bibliography includes 2,485 separate citations.

Although this book is on the whole an excellent one, it has some definite weaknesses. For example, although a description is given of the Byring, *et al.*, quasi-equilibrium rate theory of the origin of mass spectra, no use of this theory is made in the discussion of the various aspects of mass spectra such as the origin of ions of non-integral massto-charge ratio, isotope effects on fragmentation patterns, or the general nature of fragmentation patterns. A lack of understanding by the author of some important aspects of mass spectrometry is indicated by the erroneous explanation given on page 277 for the experimental data shown in Figure 124 on page 278.

A lack that will detract from the utility of this book to many analytical chemists is the very little space devoted to the important problem of the use of mass spectrometers in type or group analysis and the inadequate discussion of the use of low energy ionizing electrons in simplifying the interpretation of mass spectra and particularly in the distinction between doubly charged ions and singly charged ions of half the mass (see pages 56 and 57). On the basis of the reviewer's personal experience,

On the basis of the reviewer's personal experience, typographical errors are to be expected in a volume of the size of the present one, and they do occur. One that might be confusing to the reader inexperienced in the field occurs on page 254, where the equation describing the reaction that results in the broad peak at m/q = 30.5 in the butane mass spectra, the products are given as $(C_3H_8^+) + CH_2$ instead of the correct $C_8H_8^+ + CH_4$. As a final critical remark it should be noted that like all

As a final critical remark it should be noted that like all experts in a particular analytical technique, Beynon occasionally falls prey to the disease of over-enthusiasm. Thus he devotes two-thirds of page 344 to a description of an indirect mass spectral technique for distinguishing between a saturated hydrocarbon with four fused rings and an alkyl benzene of the same empirical formula ($C_{16}H_{26}$ not $C_{16}H_{16}$ as given in the text). As Beynon points out elsewhere in his book, any laboratory that can afford a \$100,000 mass spectrometer will have optical spectrophotometers and with an ultraviolet instrument available the solution of this problem is trivial.

The typography and the subject index are excellent, and the binding and paper appear to be of a quality that will permit the frequent use that will be given this book by practicing analytical mass spectroscopists.

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Solid State Physics. Advances in Research and Applications. Volume 11. Editors, FREDERICK SEITZ, Department of Physics, University of Illinois, Urbana, Illinois, and DAVID TURNBULL, General Electric Research Laboratory, Schenectady, New York. Academic Press Inc., 111 Fifth Avenue, New York 3, N. Y. 1960. xvi + 438 pp. 16 × 23.5 cm. Price, \$12.50.

The scientist who wishes to keep abreast of developments in the rapidly expanding field of solid state physics has been well served by the series of volumes bearing the above title. In the eleventh volume, as in the past, each article is written by an authoritative worker in the field, thereby maintaining the high standards set by the previous editions. A welcome additional feature of this volume is the inclusion of a cumulative index to the first ten volumes in the series.

The theme of a large portion of this volume is the deternuination of the detailed electronic energy band structures of solids, by galvanomagnetic, piezoresistive, photoelectronic and cyclotron resonance techniques, all of which may be classified as advances in research. This reviewer would like to see more articles dedicated to advances in applications, in keeping with the original concepts of the series.

tions, in keeping with the original concepts of the series. In the section entitled "Seniconducting Properties of Gray Tin," G. A. Busch and R. Kern review not only the solid state electronic properties of gray (or α) tin, but also the kinetics of the transition between the α - and β -modifications. The results of a large number of measurements of conductivity, Hall coefficient, magnetoresistance, thermoelectric power, magnetic susceptibility and photoconductivity are discussed, and from them the authors have chosen values of the semiconductor parameters which are thought to be most reliable. The wide variations in the data may be attributed to the difficulty of preparing pure samples.